

Nitrogen and Deuterium Hyperfine Structure in the Rotational Spectra of Pyridine and [4-D]Pyridine

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We reanalysed the ^{14}N hyperfine structure in the rotational spectra of pyridine and [4-D]-pyridine with higher precision and determined the D quadrupole coupling. It is intended to provide a reference for comparison with substituted pyridines.

We investigated by microwave Fourier transform (MWFT) [1] spectroscopy the rotational spectra of a number of substituted pyridines. The aim was a study of substitutional effects reflected by the ^{14}N quadrupole coupling. Therefore, high precision was necessary to analyse the hyperfine structure (hfs) induced by the ^{14}N nucleus and other nuclei with quadrupole moments.

To base the investigation on comparable experimental methods and analyses we remeasured as reference pyridine, $\text{C}_5\text{H}_5\text{N}$, itself, which has been the subject of many investigations, the last by Sørensen et al. [2, 3]. In this publication we present the results for pyridine and [4-D]pyridine [3, 4], which is the molecule with the least substitutional effect. The data for the other substituted pyridines (4-chloropyridine, 4-cyanopyridine, and perfluoropyridine) will be given in forthcoming publications together with a discussion of the quadrupole coupling constants.

We used our MWFT spectrometers in the range from 5.3 to 26.4 GHz described elsewhere [5–8].

Dry pyridine, 99.5%, was purchased from Merck, Darmstadt, and used without further treatment.

[4-D]Pyridine was in our stock for more than 14 years. It was purchased from Merck, Sharp and Dohme, Canada.

The spectra of both of the pyridines were recorded at temperatures of -25°C to -45°C and pressures below 0.2 Pa (1.5 m Torr). The measurements are given in Tables 1a and 1b. The frequencies given are those resulting from line shape simu-

lation [9] to account for interference of neighbouring lines. A sample of the measurements is given in Fig. 1 showing the nitrogen and deuterium hfs of the $4_{41} - 4_{22}$ transition of [4-D]pyridine.

A first order hfs analysis [10] proved to be sufficient. The rotational constants were taken from [3]. They are given with the determined coupling constants χ_{gg} , $g = a, b, c$ in Table 2. For [4-D]pyridine a coupled basis: $F_1 = J + I$ (nitrogen), $F = F_1 + I$ (deuterium), J angular momentum and I spin operator, was used. For pyridine the analysis was made with the programs HT1NQ and DH14KS [11] and for [4-D]pyridine with Q2SIM [12] and Q2FIT [13].

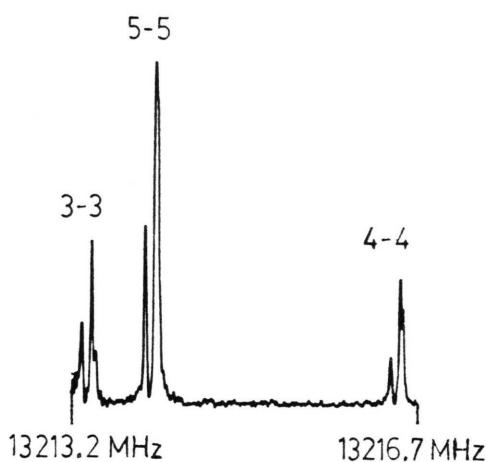


Fig. 1. Rotational transition $J_{K_a, K_c} = 4_{41} - 4_{22}$ of [4-D]pyridine showing ^{14}N - and D-hfs. Range from 13213.2 to 13216.7 MHz out of a 10 MHz recording. Temperature -35°C , pressure 0.2 mTorr, polarizing frequency 13213.2 MHz, 50 ns sampling interval, 10^7 averaging cycles, 1024 data points supplemented with 3072 zeros prior to Fourier transformation. Spectral point distance 2.5 kHz. F_1 quantum numbers are above the lines.

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$\Delta\nu$ hfs-splittings referred to the strongest multiplet component, ν_0 hypothetical unsplit lines, $\Delta(\Delta\nu) = \Delta\nu - \Delta\nu_{\text{calc}}$ deviations of the splittings, $\Delta\nu_{\text{calc}}$ splittings calculated with the constants of Table 2, $\Delta\nu_0$ deviations from rigid rotor spectrum calculated with the rotational constants of Table 2 (program DH9 [11]).

J'	K'_L	$K'_L - J''$	K''	K''_+	$F' - F''$	ν [MHz]	$\Delta\nu$ [MHz]	ν_0 [MHz]	$\Delta(\Delta\nu)$ [kHz]	$\Delta\nu_0$ [kHz]	J'	K'_L	$K'_L - J''$	K''	K''_+	$F' - F''$	ν [MHz]	$\Delta\nu$ [MHz]	ν_0 [MHz]	$\Delta(\Delta\nu)$ [kHz]	$\Delta\nu_0$ [kHz]
1	0	1 - 0	0	0	1-1	8762.887	-1.477	8764.118	-5	-5	5	4	1 - 5	4	2	5-5	7193.367	-0.654	7193.839	+2	-209
					2-1	8764.364			-1							6-6	7194.021				
					0-1	8766.572	+2.208		-1							4-4	7194.160	+0.139		+5	
2	0	2 - 1	0	1	1-0	14901.463	-1.544	14902.987	+1	-15	5	5	1 - 5	3	2	4-4	11162.700	-0.416	11163.689	+4	-155
					2-2	14901.810	-1.197		-2							6-6	11163.116				
					3-2	14903.007			-9							5-5	11165.178	+2.062		+1	
					2-1	14903.276	+0.269		+3												
					1-1	14905.146	+2.139														
2	1	1 - 1	1	0	2-1	20372.681	-1.502	20373.907	+1	-39	6	4	2 - 6	4	3	6-6	14149.272	-0.104	14149.346	+2	-470
					1-1	20373.399	-0.784		+2							7-7	14149.376			-4	
					2-2	20373.736	-0.447		+14							5-5	14149.390	+0.014			
					3-2	20374.183										6-6	6553.758	-0.724		+5	
					1-0	20376.003	+1.820		0							7-7	6554.482			-1	
																5-5	6554.606	+0.124		+1	-306
2	1	2 - 1	1	1	2-1	14681.302	-1.549	14682.531	-2	-16	6	6	1 - 6	4	2	5-5	12340.275	-0.363	12341.261	+5	-200
					2-2	14681.728	-1.123		-6							7-7	12340.638				
					3-2	14682.851			-8							6-6	12342.822	+2.184		+4	
					1-1	14683.033	+0.182		-6												
					1-0	14684.111	+1.260														
2	1	1 - 2	1	2	2-3	8536.463	-0.466	8537.075	-5	-25	7	5	2 - 7	5	3	7-7	13687.498	-0.334	13687.735	-1	-682
					1-1	8536.563	-0.366		-1							8-8	13687.832			+1	
					3-3	8536.929			-1							6-6	13687.881	+0.049			
					2-2	8537.584	+0.655		-6							7-7	5841.363	-0.744		+6	
					3-2	8538.040	+1.111		-5							8-8	5842.107			-2	
					1-2	8538.297	+1.368		-5							6-6	5842.217	+0.110		+2	
2	2	1 - 2	0	2	1-1	9251.818	-1.544	9253.978	-4	-26	8	6	2 - 8	6	3	8-8	13069.681	-0.502	13070.034	0	-942
					3-3	9253.362			-12							9-9	13070.183			-3	
					1-2	9253.668	+0.306		-1							7-7	13070.243	+0.060			
					2-1	9254.275	+0.913		-1												
					3-2	9254.556	+1.194		-5							8-8	5078.677	-0.733		-2	
					2-3	9254.932	+1.570		+1							9-9	5079.410			-4	
					2-2	9256.135	+2.773									7-7	5079.498	+0.088			
3	1	2 - 3	1	3	2-2	14752.846	-0.444	14753.608	+1	-75	9	7	2 - 9	7	3	9-9	12300.998	-0.621	12301.433	-1	-1251
					4-4	14753.280			+4							10-10	12301.619			+1	
					3-3	14754.566	+1.276									8-8	12301.689	+0.070			
3	2	1 - 3	2	2	3-3	8194.350	-0.161	8194.471	+1	-69	10	8	2 - 10	8	3	10-10	11397.481	-0.697	11397.967	-2	-1603
					4-4	8194.511			-3							11-11	11398.178				
					2-2	8194.565	+0.054		-3							9-9	11398.244	+0.066		-4	
3	3	1 - 3	1	2	2-2	9667.811	-0.755	9669.104	-3	-62	11	9	2 - 11	9	3	11-11	10382.130	-0.736	10382.641	-1	-1996
					4-4	9668.566			+1							12-12	10382.866			-4	
					3-3	9670.717	+2.151									10-10	10382.929	+0.063			
4	2	2 - 4	2	3	3-3	14654.305	-0.151	14654.618	+5	-175	12	10	2 - 12	10	3	12-12	9282.599	-0.745	9283.115	0	-2410
					5-5	14654.456			+1							13-13	9283.344			-1	
					4-4	14655.065	+0.609		-3							11-11	9283.405	+0.061			
4	3	1 - 4	3	2	4-4	7743.927	-0.500	7744.294	-3	-129	13	11	2 - 13	11	3	13-13	8130.497	-0.730	8131.001	-1	-2820
					5-5	7744.427			-6							14-14	8131.227			-1	
					3-3	7744.549	+0.122									12-12	8131.282	+0.055			
4	4	1 - 4	2	2	3-3	10288.149	-0.516	10289.203	+2	-108	18	15	3 - 18	15	4	18-18	11490.952	-0.711	11491.439	-2	-8866
					5-5	10288.665			+3							19-19	11491.663			+1	
					4-4	10290.684	+2.019		-9							17-17	11491.703	+0.040			
5	3	2 - 5	3	3	4-4	14463.827	-0.048	14463.929	-9	-297	19	16	3 - 19	16	4	19-19	9964.828	-0.684	9965.295	0	-9759
					6-6	14463.875			-9							20-20	9965.512			-2	
					5-5	14464.059	+0.184									18-18	9965.546	+0.034			
																20-20	8455.586	-0.641	8456.424	-2	-10439
																21-21	8455.627			-3	
																19-19	8456.656	+0.029			

Table 1 b. Measured rotational transitions of [4-D]pyridine showing ^{14}N (F_1 quantum numbers) and D hyperfine structure (F quantum numbers). See also Table 1 a.

J'	K'_-	$K'_+ - J''$	K''_-	K''_+	$F' - F'_-$	$F' - F''$	ν [MHz]	$\Delta\nu$ [MHz]	ν_0 [MHz]	$\Delta(A\nu)$ [kHz]	$\Delta\nu_0$ [kHz]	J'	K'_-	$K'_+ - J''$	K''_-	K''_+	$F' - F'_-$	$F' - F''$	ν [MHz]	$\Delta\nu$ [MHz]	ν_0 [MHz]	$\Delta(A\nu)$ [kHz]	$\Delta\nu_0$ [kHz]		
1	0	1	-	0	0	0	1-1, 1-0 1-1, 1-1 1-1, 1-2 1-1, 2-1 1-1, 2-2 1-1, 0-1 2-1, 1-0 2-1, 1-1 2-1, 3-2 2-1, 2-1 2-1, 2-2 0-1, 1-1 0-1, 1-2	8274.637 8274.668 8274.711 8276.107 8276.129 8276.173 8278.345	-1.492 -1.461 -1.418 -0.022 8275.894 +0.044 +2.216	-	3 3 4 4 - 5	3	3	3	1	-	3	1	2	2-2, 2-2 2-2, 3-3 2-2, 1-1 4-4, 4-4 4-4, 5-5 4-4, 3-3 3-3, 3-3 3-3, 4-4 3-3, 2-2	11013.860 11013.937 11013.974 11014.701 11014.798 11014.823 11017.185 11017.260 11017.291	-0.938 -0.861 -0.824 -0.097 11015.389 +0.025 +2.387 +2.462 +2.493	+	9 9 4 4 - 2 3 4 9	65
2	0	2	-	1	0	1	1-0, 1-1 1-0, 2-1 3-2, 3-2 3-2, 4-3 3-2, 2-1 2-1, 1-0 2-1, 3-2 2-1, 2-1 1-1, 2-2	14505.901 14507.377 14507.48 14507.52 14507.58 14509.576	-1.476 14507.331 +0.103 +0.143 +0.203 +2.199	+	1 - 3	8 9 1 6	4	3	2	-	4	1	3	3-3, 3-3 3-3, 4-4 3-3, 2-2 5-5, 5-5 5-5, 6-6 5-5, 4-4 4-4, 4-4 4-4, 5-5 4-4, 3-3	14744.174 14744.215 14744.461 14744.502 14745.603 14745.639	-0.328 -0.287 -0.041 14744.790 +1.101 +1.137	+	3 6 1 - 2 3	136
2	1	2	-	1	1	1	2-1, 1-1 2-1, 1-0 2-1, 3-2 2-1, 2-1 3-2, 2-1 3-2, 4-3 3-2, 3-2 1-1, 2-2 1-0, 1-1 1-0, 2-1	13986.304 13987.836 13987.892 13988.018 13989.113	-1.532 13987.535 +0.056 +0.182 +1.277	-	1 7	22 13	5	4	2	-	5	2	3	3-3, 3-3 3-3, 4-4 5-5, 5-5 5-5, 6-6 5-5, 4-4 4-4, 4-4 4-4, 5-5 4-4, 3-3	13213.311 13213.401 13213.440 13213.943 13214.045 13214.065 13215.441 13216.528 13216.552	-0.734 -0.644 -0.605 -0.102 13214.682 +0.020 +2.396 +2.483 +2.507	+	1 1 6 3 - 2 5 4	118
2	2	1	-	2	0	2	1-1, 1-2 1-1, 2-2 1-1, 2-1 3-3, 3-3 3-3, 4-4 3-3, 2-2 1-2, 2-3 2-3, 3-4 2-2, 2-2 2-2, 3-3 2-2, 1-1	9646.384 9647.916 9648.007 9648.042 9648.425 9649.575 9650.837 9650.892 9650.918	-1.623 -0.091 9648.623 +0.035 +0.418 +1.568 +2.830 +2.885 +2.911	+	8 4 3 2 - 6 8 4	30	5	5	1	-	5	3	2	4-4, 4-4 4-4, 5-5 4-4, 3-3 6-6, 6-6 6-6, 7-7 6-6, 5-5 5-5, 5-5 5-5, 6-6 5-5, 4-4	15412.473 15412.51 15412.53 15412.689 15412.732 15413.769 15413.807	-0.259 -0.222 -0.202 -0.043 15413.016 +1.037 +1.075	+	2 2 12 0 - 3 1	214
5	5	1	-	5	3	2	4-4, 4-4 4-4, 5-5 4-4, 3-3 6-6, 6-6 6-6, 7-7 6-6, 5-5 5-5, 5-5 5-5, 6-6 5-5, 4-4	16374.937 16375.050 16375.076 16375.477 16375.583 16375.583 16378.08 16378.187 16378.208	-0.646 -0.533 -0.507 -0.106 16376.270 +2.497 +2.604 +2.625	+	1 10 9 13 - 4 11 10	210													
6	5	2	-	6	3	3	5-5, 5-5 5-5, 6-6 5-5, 4-4 5-5, 6-6 5-5, 5-5 5-5, 6-6 7-7, 7-7 7-7, 8-8 7-7, 6-6 6-6, 6-6 6-6, 7-7 6-6, 5-5	16658.099 16658.155 16658.295 16658.347 16658.486 16659.532	-0.248 -0.192 -0.052 16658.668 +1.139 +1.185	+	2 8 2 - 2 2 0	285													

Table 2. Rotational and quadrupole coupling constants [MHz] of pyridine and [4-D]pyridine. Assumptions of the analysis in square brackets. σ standard deviation of the fit in kHz, $|\chi_+, \chi_-|$ maximum correlation coefficient. Single standard errors in brackets. Derived parameters below the line.

Pyridine		[4-D]Pyridine	
<i>A</i>	[6039.2516(6)][3]	<i>A</i>	[6038.9967(10)][3]
<i>B</i>	[5804.9116(6)]	<i>B</i>	[5420.0697(9)]
<i>C</i>	[2959.2117(6)]	<i>C</i>	[2855.8194(8)]
$\chi_+ = \chi_{bb} + \chi_{cc}$	4.908(3)	$\chi_{aa} (^{14}\text{N})$	-4.907(3)
$\chi_- = \chi_{bb} - \chi_{cc}$	-2.041(5)	$\chi_{bb} (^{14}\text{N})$	1.442(5)
		$\chi_{aa} (\text{D})$	0.196(4)
		$\chi_{bb} (\text{D})$	-0.090(8)
σ	4	σ	6
$ \chi_+, \chi_- $	0.13	$ \chi_{aa} (\text{N}), \chi_{bb} (\text{N}) $	0.22
χ_{aa}	-4.908(3)	$\chi_{cc} (^{14}\text{N})$	3.465(5)
χ_{bb}	1.434(3)		
χ_{cc}	3.474(3)	$\chi_{cc} (\text{D})$	-0.106(8)

Our values for pyridine and [4-D]pyridine show standard errors which are by a factor 3 less than those from previous measurements [2, 4]. The ^{14}N quadrupole coupling constants of the two isotopomers agree roughly within the error limits. The D quadrupole coupling constants agree with values calculated with an ab initio SCF method by Huber [14].

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